

First-principles prediction of structural and electronic properties of BiVO_3

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Bismuth vanadate, whose chemical formula is BiVO_4 , has been attracted attention for sustainable energy applications since this material shows promising results for solar hydrogen production when it is used as water-splitting photocatalyst and photoanode in photoelectrochemical cell [1]. In recent years, the ABO_3 -type perovskites and related heterostructures have provided a variety of physical properties such as ferroelectricity, ferromagnetism, and multiferroicity [2]. From this perspective, it is wondering what properties could be generated if bismuth vanadate is materialized as BiVO_3 . I thus attempt to predict the structural and electronic properties of BiVO_3 using first-principles calculations. Those properties are not yet reported by experiment since the synthesis of BiVO_3 is very difficult using conventional growth techniques such as solid-state reaction method.

In this work, first-principles calculations are based on density-functional theory and the screened hybrid functional of Heyd-Scuseria-Ernzerhof (HSE) [3,4], implemented with the projector augmented-wave method in the vasp code [5]. The mixing parameter in the HSE (i.e., amount of the nonlocal Fock-exchange) was set to 25%. I have performed bulk calculations for BiVO_3 and nominal form, BiVO_4 is also considered for comparison. Two BiVO_4 phases (monoclinic: $I2/b$ [convertible to a space group of $C2/c$ with a body-centered cell] and tetragonal structure: $I4_1/a$) and BiVO_3 (cubic perovskite: $Pm3m$) are studied.

Figure 1 shows crystal structures of BiVO_4 and BiVO_3 . The unitcells contain 20 atoms for BiVO_4 and five atoms for BiVO_3 . The calculated lattice parameters of monoclinic BiVO_4 are $a = 5.09 \text{ \AA}$, $b = 5.15 \text{ \AA}$, and $c = 11.66 \text{ \AA}$, and those of tetragonal form are $a = 7.32 \text{ \AA}$ and $c = 6.43 \text{ \AA}$. These HSE results are in good agreement with the experimental values. For the cubic BiVO_3 , the lattice constant is computed to be 3.85 \AA . Moving to the electronic structure of BiVO_3 , which is a main concern, our calculations predict the metallic-like band structure [Figure 2], resulting from two extra electrons that do not participate in chemical bonds. When the electrons are intentionally removed in the calculations, BiVO_3 exhibits an insulating behavior with the band gap of 2.9 eV at X-point. It is noteworthy that the ferromagnetic phase of BiVO_3 is more stable than the simple metallic state of it by 0.8 eV per formula unit. Therefore, various magnetic phases will be examined in a following study.

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References

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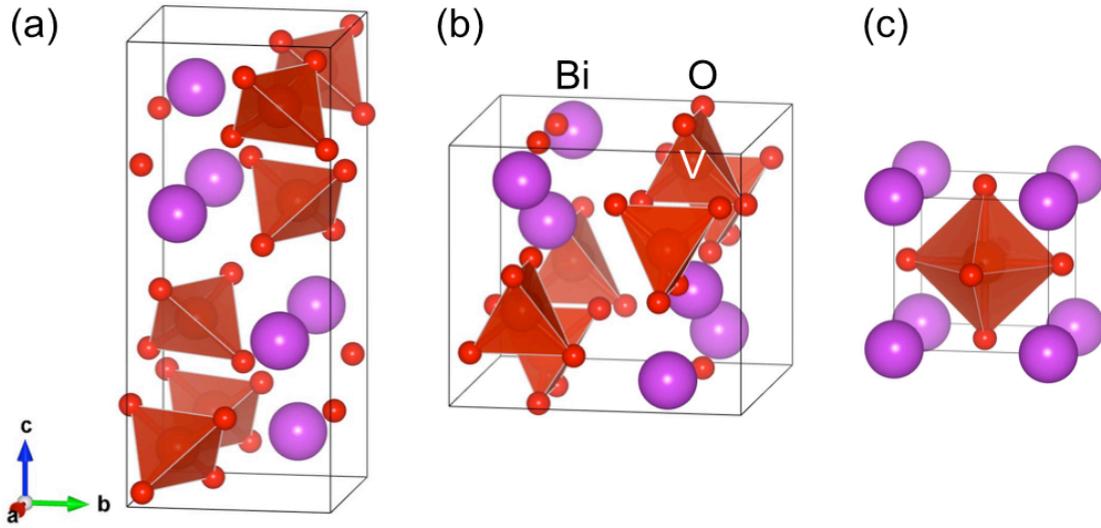


FIG. 1. Crystal structures of (a) monoclinic BiVO_4 , (b) tetragonal BiVO_4 , and (c) cubic perovskite BiVO_3 . Magenta and red balls denote bismuth and oxygen atoms, respectively. Ball in polyhedrals is vanadium.

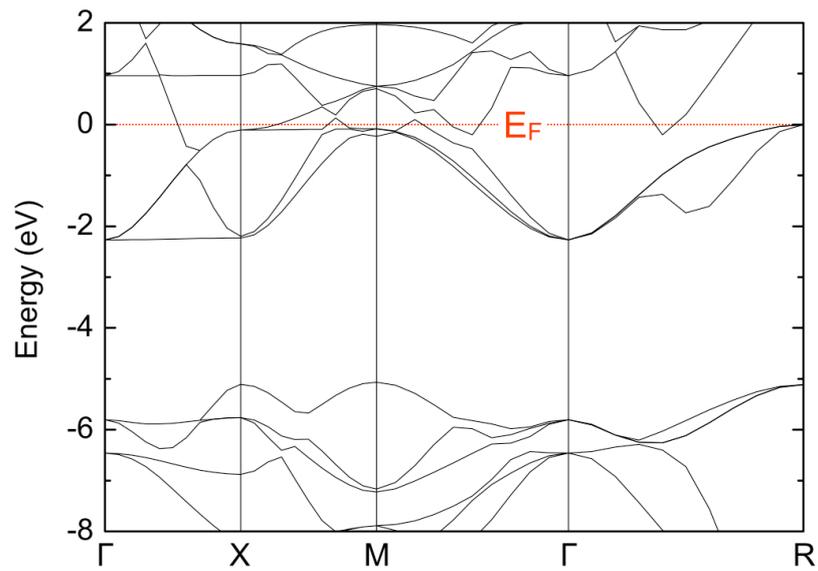


FIG. 2. Predicted band structure of cubic BiVO_3 using the HSE functional. The Fermi level (E_F) is set as zero.